PS L2 Syllabus Review Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

1. What is the name of this course?
2. What is the name of your teacher?
3. List three of your five required daily materials:
	*
	*
	*
4. Identify and describe one of this course’s ethics:

PS L2 Syllabus Review Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

1. What is the name of this course?
2. What is the name of your teacher?
3. List three of your five required daily materials:
*
*
*
1. Identify and describe one of this course’s ethics:

PS L4 Do Now! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Identify 4 items or concepts that you missed on the Ch. 02 Pre-Test:

*
*
*
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L4 Do Now! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Identify 4 items or concepts that you missed on the Ch. 02 Pre-Test:

*
*
*

PS L5 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

1. Define matter:
2. Define chemistry:
3. Define pure substance:
4. A substance that cannot be broken down into simpler substances by ordinary chemical means is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
5. The smallest unit of an **element** that keeps the **element’s chemical properties** is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
6. The smallest unit of a **substance** that behaves like the **substance** is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
7. Which seven (7) elements exist as diatomic molecules?

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
8. Which elements exist as a tetratomic molecule? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
9. Which elements exist as an octatomic molecule? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
10. What is the symbol for:

	1. Hydrogen \_\_\_\_\_\_\_\_\_\_
	2. Helium \_\_\_\_\_\_\_\_\_\_
	3. Lithium \_\_\_\_\_\_\_\_\_\_
	4. Beryllium \_\_\_\_\_\_\_\_\_\_
	5. Boron \_\_\_\_\_\_\_\_\_\_
	6. Carbon \_\_\_\_\_\_\_\_\_\_
	7. Nitrogen \_\_\_\_\_\_\_\_\_\_
	8. Oxygen \_\_\_\_\_\_\_\_\_\_
	9. Fluorine \_\_\_\_\_\_\_\_\_\_
	10. Neon \_\_\_\_\_\_\_\_\_\_

PS L6 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Write down the names and symbols of the elements on Elements Quiz 1 – 10.

PS L6 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Write down the names and symbols of the elements on Elements Quiz 1 – 10.

PS L7 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

1. A substance that is made from two or more chemically combined elements is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
2. APPLICATION: Identify the following as an element (E) or compound (C).
	1. \_\_\_\_\_\_\_ He
	2. \_\_\_\_\_\_\_ H2O
	3. \_\_\_\_\_\_\_ CO
	4. \_\_\_\_\_\_\_ Na
	5. \_\_\_\_\_\_\_ CH4
	6. \_\_\_\_\_\_\_ CO2
	7. \_\_\_\_\_\_\_ Sn
	8. \_\_\_\_\_\_\_ Cl2
3. The properties of a compound is (similar to / different than) the properties of the elements that they are made from.
4. A combination of substances that are not chemically combined is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
5. When a substance is “able to be mixed”, it is \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
6. When the components are evenly distributed in a mixture, it is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
7. When the mixture is different throughout, it is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
8. When the substances are not evenly distributed in a mixture, it is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
9. When the mixture is the same throughout, it is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
10. APPLICATION: Identify the following as a homogeneous (HOM) or a heterogeneous (HET) mixture.
	1. \_\_\_\_\_\_\_ Apple juice
	2. \_\_\_\_\_\_\_ Italian Salad
	Dressing
	3. \_\_\_\_\_\_\_ The ocean
	4. \_\_\_\_\_\_\_ Atmosphere
	5. \_\_\_\_\_\_\_ Coca-Cola
	6. \_\_\_\_\_\_\_ The Earth
	7. \_\_\_\_\_\_\_ Ground Cinnamon
	8. \_\_\_\_\_\_\_ Cardamom Seed
11. Complete the following chart:

|  |  |  |
| --- | --- | --- |
|  | COMPOUND | MIXTURE |
| Combination |  |  |
| Properties |  |  |
| Composition |  |  |

PS L8 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Review of Previous Material

1. Define chemistry:
2. The smallest unit of an **element** that keeps the **element’s chemical properties** is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
3. Define pure substance:
4. A substance that cannot be broken down into simpler substances by ordinary chemical means is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
5. Define matter:
6. The smallest unit of a **substance** that behaves like the **substance** is a(n) \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.
7. Which seven (7) elements exist as diatomic molecules?

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
8. Which element exists as a tetratomic molecule? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_
9. Which element exists as an octatomic molecule? \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

PS L9 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Using the \*words listed on the **Section 2.1 Spider Web Concept Map**, rearrange and connect the words using arrows and **true statements**. For example:

 classified classified

 Pure Substance Matter Mixture

 as a as a

\*Feel free to add your own words

PS L10 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Using the following words pairs, write a **true** statement.

1. Chemistry + Matter:
2. Pure Substance + Composition:
3. Element + Atom:
4. Compound + Elements:
5. Mixture + Properties:

Fill in the symbols for the first 30 elements:

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PS L11 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Complete Section 2.1 Review on Page 50.

PS L12 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Define physical property:

List some examples of physical properties:

Physical properties can help \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ substances, can be \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ and \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_, and help determine \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.

Pick any (non-human) object in the room, name it, and provide 5 different physical properties.

Object:

5 Physical Properties:

PS L12 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Define physical property:

List some examples of physical properties:

Physical properties can help \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ substances, can be \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_ and \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_, and help determine \_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_.

Pick any (non-human) object in the room, name it, and provide 5 different physical properties.

Object:

5 Physical Properties:

PS L13 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Identify each substance as a pure substance (PS) or mixture (MIX), and then identify each as an element (E), compound (C), homogeneous mixture (HOM), or heterogeneous mixture (HET).

|  |  |  |
| --- | --- | --- |
| **Identity of Substance** | **Pure Substance (PS) or Mixture? (MIX)** | **Element (E),** **Compound (C),** **Homogeneous Mixture (HOM), or****Heterogeneous Mixture (HET)?** |
| 1. Nitrogen Gas (N2) |  |  |
| 2. Deer Park Water |  |  |
| 3. Red Delicious Apple |  |  |
| 4. 24K (Pure) Gold (Au) |  |  |
| 5. A&W® Root Beer |  |  |
| 6. Chocolate Chip Cookie |  |  |
| 7. Heinz® Ketchup |  |  |
| 8. Hydrogen Peroxide (H2O2) |  |  |
| 9. Sodium Hydroxide (NaOH)  |  |  |
| 10. Apple iPhone® |  |  |

Fill in the symbols for the first 48 elements in their **correct** location:

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PS L14 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Complete Section 2.2 Review on Page 58: 1 – 6. (Graded as HW)

PS L15 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Complete Section 2.3 Review on Page 64. (Graded as HW)

PS L16 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Complete Chapter 2 Review on Page 70: 1-14

PS L17 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Complete the Preview of Section 4.1 . . .

PS L18 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Fill in the symbols for the first 79 elements in their **correct** location:

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PS L18 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Fill in the symbols for the first 79 elements in their **correct** location:

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PS L19 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Fill in the symbols for the first 92 elements in their **correct** location:

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What was Democritus’s contribution to the atomic model?

What was the alchemist’s contribution to the atomic model?

PS L20 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Fill in the symbols for the first 92 elements in their **correct** location:

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What was Boyle’s contribution to the atomic model?

What was Dalton’s contribution to the atomic model? (He had five!)

How did JJ Thomson discover the electron?

What was Lord Kelvin’s contribution to the atomic model?

PS L21 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Why was it challenging to determine the shape of the hidden object in the Rutherford’s Atoms Lab

PS L22 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Fill in the symbols for the first 92 elements in their **correct** location:

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What was Rutherford’s contribution to the atomic model?

PS L23 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Part I: Fill in the symbols for the first 92 elements in their **correct** location:

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Part II: Complete the **Cumulative Element Practice #1**

PS L24 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Part I: Fill in the symbols for the first 92 elements in their **correct** location:

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Part II: Complete the Cumulative Element Practice #2.

PS L25 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

TBD: Atomic Math; Section 4.2 Notes; **Review of Year**?

What are the 3 steps of fence-posting?

1.
2.

PS L26 DO NOW! Name:\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_\_

Part I: Fill in the symbols for the first 92 elements in their **correct** location:

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Part II: Complete the Cumulative Element Practice #3: 1 – 92 (Selected)